

## Seasonal variability of the effect of coral reefs on seawater CO<sub>2</sub> and air–sea CO<sub>2</sub> exchange

Nicholas R. Bates<sup>1</sup>

Bermuda Biological Station For Research (BBSR), Inc., 17 Biological Station Lane, Ferry Reach, GEO1, Bermuda

### Abstract

There are complex physical and biological processes controlling the exchange of carbon dioxide (CO<sub>2</sub>) between the ocean and atmosphere. In coral reef ecosystems, the balance of biological processes such as calcium carbonate (CaCO<sub>3</sub>) formation and organic carbon production can either lead to CO<sub>2</sub> being retained in the oceanic environment (i.e., oceanic sink of CO<sub>2</sub>) or returned to the atmosphere through gas exchange (oceanic source of CO<sub>2</sub>). What remains uncertain is the fate of CO<sub>2</sub> in reefs subject to seasonal change and the annual balance of air–sea CO<sub>2</sub> flux in such systems. Here it is shown that the Bermuda coral reef acts as a source of CO<sub>2</sub> to seawater overlying the reef. The magnitude of this source of CO<sub>2</sub> varies seasonally in response to changes in the reef community between coral- and macroalga-dominated states, reflecting changes in the net balance between calcification and organic carbon production. With knowledge of the calcification rate (~5.6 to 10.6 g CaCO<sub>3</sub> m<sup>-2</sup> d<sup>-1</sup>) and observed modification in seawater *f*CO<sub>2</sub> by reef metabolism, rates (-0.6 to 3.3 g C m<sup>-2</sup> d<sup>-1</sup>) and seasonal patterns of macroalgal productivity were estimated. Whether the Bermuda coral reef system acts as an oceanic sink or source of CO<sub>2</sub> to the atmosphere not only depends on this seasonal variation, but, more importantly, depends on the pre-existing air–sea CO<sub>2</sub> disequilibrium of open ocean waters surrounding the reef system. The Bermuda coral reef system serves as a useful model for understanding the fate of CO<sub>2</sub> in other reefs, particularly those reefs changing because of environmental stress.

Coral reefs are biologically diverse ecosystems built upon the accumulation of calcium carbonate (CaCO<sub>3</sub>) produced by frame-building scleractinian corals and calcareous algae. Within the reef system, biological processes such as photosynthesis (i.e., CO<sub>2</sub> + H<sub>2</sub>O ⇌ CH<sub>2</sub>O + O<sub>2</sub>) and calcification (i.e., Ca<sup>2+</sup> + CO<sub>3</sub><sup>2-</sup> = CaCO<sub>3</sub>) have demonstrable effects on seawater CO<sub>2</sub> chemistry (e.g., Langdon et al. 2000). For example, photosynthetic fixation of CO<sub>2</sub> into organic carbon acts to decrease the fugacity of CO<sub>2</sub> (*f*CO<sub>2</sub>) and total carbon dioxide (TCO<sub>2</sub>, where TCO<sub>2</sub> = HCO<sub>3</sub><sup>-</sup> + CO<sub>3</sub><sup>2-</sup> + CO<sub>2</sub>), respectively, whereas total alkalinity (TA, where TA = HCO<sub>3</sub><sup>-</sup> + 2CO<sub>3</sub><sup>2-</sup> + B(OH)<sub>4</sub><sup>-</sup> + OH<sup>-</sup> + H<sup>+</sup> + minor species) remains largely unaffected. Calcification (Ca + 2HCO<sub>3</sub><sup>-</sup> ⇌ CaCO<sub>3</sub> + CO<sub>2</sub> + H<sub>2</sub>O) acts to decrease TCO<sub>2</sub> and total alkalinity in the ratio of 1:2. The fate of CO<sub>2</sub> produced by calcification depends on many factors, including the ratio of organic carbon production to CaCO<sub>3</sub> production (Budde-meier 1996), irradiance levels (Gattuso et al. 1997), and the dominant type of calcifying organism (Gattuso et al. 1999a).

Several studies have demonstrated that coral-dominated reef ecosystems are sources of CO<sub>2</sub> to the surrounding waters (e.g., Ware et al. 1992; Gattuso et al. 1993a, 1995, 1996a,

1999b). Other studies have suggested that macroalga-dominated reef systems are sinks for CO<sub>2</sub> (e.g., Yamamuro et al. 1995; Kayanne et al. 1996; Gattuso et al. 1997; Kraines et al. 1997). In a recent review, Gattuso et al. (1999b) find that “average” coral reef flats are sources of CO<sub>2</sub> to the atmosphere. Whether coral reef ecosystems act as sources of CO<sub>2</sub> to the atmosphere or sinks of CO<sub>2</sub> appears to depend on the balance of two processes: organic carbon production by macroalgae and CaCO<sub>3</sub> production by corals. This balance is critical for carbon cycling on reefs because many reefs are shifting from coral-dominated to macroalga-dominated states in response to environmental change (e.g., Kinsey 1985; Smith and Buddemeier 1992; Hughes 1994), thereby potentially shifting reef metabolism from CO<sub>2</sub> release through reef calcification to CO<sub>2</sub> invasion driven by organic carbon production. Furthermore, it is postulated that coral reef calcification rates may decline in the future because of the increasing levels of atmospheric CO<sub>2</sub> change and the reduction of inorganic carbon species available for calcification (e.g., Gattuso et al. 1999a; Kleypas et al. 1999; Langdon et al. 2000).

The fate of CO<sub>2</sub> in coral reef ecosystems subject to seasonal changes and the annual balance of CO<sub>2</sub> in such systems remain uncertain. Previous studies have typically been restricted to observations of seawater CO<sub>2</sub> and associated variables over a few days only (e.g., Gattuso et al. 1993b; Kayanne et al. 1995; Frankignoulle et al. 1996; Kawahata et al. 1997; Ohde and Woesik 1999) to approximately 1 month (Bates et al. 2001). In the latter study, Bates et al. (2001) found that the Bermuda coral reef was a source of CO<sub>2</sub> to the water overlying the reef during a short period of observation. In this paper, the seasonal and interannual variability of CO<sub>2</sub> and air–sea CO<sub>2</sub> fluxes on the Bermuda coral reef over a 4-yr period (1994–1998) are investigated. The Bermuda reef CO<sub>2</sub> data are compared to contemporaneous sea-

<sup>1</sup> Corresponding author (nick@bbsr.edu).

### Acknowledgments

The author thanks the following individuals for their help and assistance: S. J. Bell, J. Benson, B. Bjork, F. Howse, F. Bahr, R. J. Johnson, M. Church, S. Becker, and R. Little (BBSR); the crew and captains of R/V *Weatherbird II*; and T. Takahashi, D. W. Chipman, and J. G. Goddard (Lamont Doherty Earth Observatory). I thank J. A. Kleypas, F. T. Mackenzie, J.-P. Gattuso, and S. R. Smith for their constructive comments on earlier versions of this manuscript. Two reviewers are thanked for their thoughtful and constructive comments. This research was supported by the National Science Foundation (OCE-9416565 and OCE-9818878 to N.R.B.).

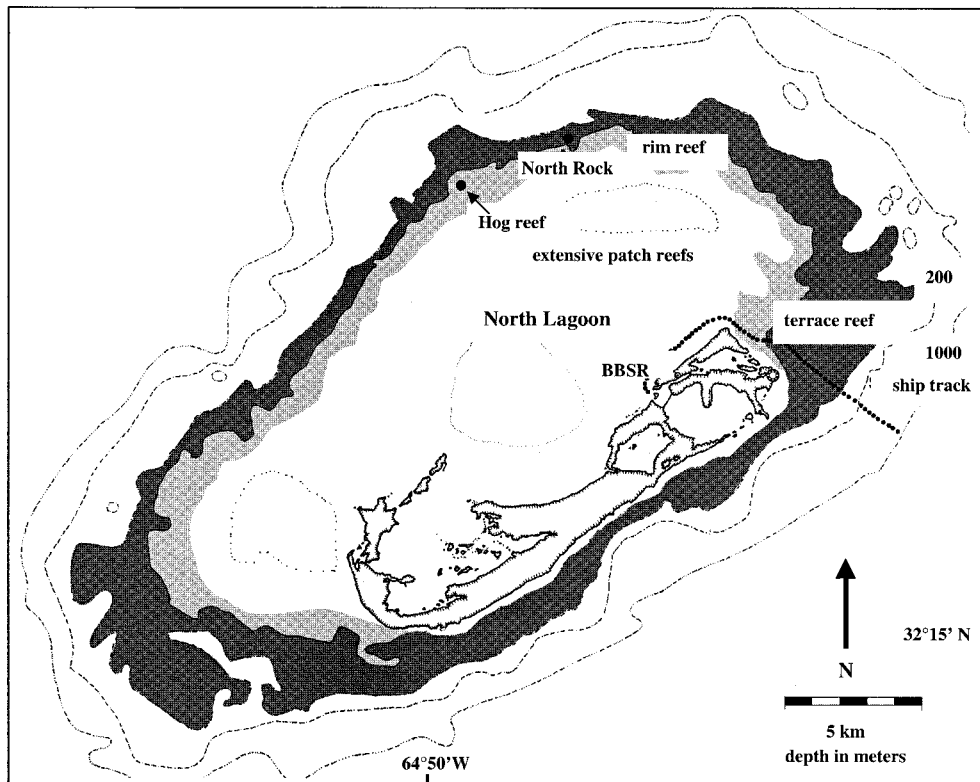


Fig. 1. Location map of the Bermuda coral reef. The route of the R/V *Weatherbird II* between the dock at BBSR and the open ocean through the eastern rim and ledge-flat reefs of Bermuda is noted on the map. The oceanographic time series site, Bermuda Atlantic Time-series Study (BATS), is located ~85 km southeast of Bermuda in a water depth of 4,500 m at 31°50'N, 64°10'W.

water CO<sub>2</sub> data collected from the surrounding offshore Sargasso Sea at the U.S. Joint Global Ocean Flux Study (JGOFS) Bermuda Atlantic Time-series Study (BATS) site (31°50'N, 64°10'W). Bermuda reef waters have short residence times and are continuously renewed from offshore water impinging on the reef system. Comparison of both CO<sub>2</sub> time series thus reveals the seasonal biogeochemical modification of Sargasso Sea water after entrainment onto the Bermuda reef. The magnitude of the Bermuda coral reef system as a source of CO<sub>2</sub> changes in response to seasonal changes in the community structure of the reef (i.e., coral-dominated vs. macroalga-dominated ecosystem) and the balance of organic carbon production to CaCO<sub>3</sub> production.

#### Materials and methods

*The Bermuda coral reef system*—The shallow platform waters of Bermuda are surrounded with a flank of outer terrace and rim reefs that define the North Lagoon (Fig. 1). The North Lagoon contains inshore and nearshore patch reefs and extensive sand areas, with an average depth of 8.5 m (Morris et al. 1977). The mean percent cover of hard corals and macroalgae in Bermuda waters is 20% and 11%, respectively (CARICOMP 1997). The percent cover of hard corals in the rim and terrace reefs is higher than the mean (~20–30%), whereas percent cover on the outer slope, in depths of 15–30 m, is ~60% (S. R. Smith unpubl. data).

Waters of the Bermuda platform are continuously exchanging with offshore waters of the surrounding Sargasso Sea. In the North Lagoon, residence times of water vary between 1 and 20 tidal cycles (Morris et al. 1977; R. J. Johnson unpubl. data); the variability of residence time relates to patterns of tidal and wind circulation and distance from offshore waters. On the southeastern flank of the Bermuda coral reef, through which the ship was routed, the residence time of water in the area is typically 1–3 tidal cycles (R. J. Johnson unpubl. data). There is no significant freshwater discharge from the Bermuda landmass into the shallow waters of the Bermuda platform; thus, the source waters of the Bermuda coral reef are the surrounding offshore waters of the Sargasso Sea.

*Seawater and atmospheric CO<sub>2</sub> sampling*—Two contemporaneous time series of CO<sub>2</sub> and associated variables were collected during a 4-yr period on the Bermuda coral reef and at an offshore open ocean site in the surrounding Sargasso Sea (i.e., at the BATS site, located ~85 km southeast of Bermuda at 32°50'N, 64°10'W), which is part of the North Atlantic subtropical gyre. Surface seawater data (i.e., *f*CO<sub>2</sub> and temperature) were collected aboard the R/V *Weatherbird II* during approximately 150 cruises between 1994 and 1998. The ship transits between the dock at the Bermuda Biological Station For Research (BBSR) and the open ocean BATS site. The ship track is routed over ~6 km of the southeastern

outer rim and terrace reefs of Bermuda (Fig. 1). Coral framework builders, such as *Montastrea annularis* and *Diploria labyrinthiformis* dominate the reef, with a hard coral cover of approximately 30%. The last few kilometers of the ship route close to BBSR passes the nearshore patch reefs and sandy biotope containing occasional beds of sea grasses (e.g., *Thalassia*) and macroalgae.

Seawater and atmospheric  $f\text{CO}_2$  measurements were collected across the Bermuda coral reef and the BATS site using a nondispersive infrared analyzer and showerhead equilibrator. Data were collected every  $\sim 2$  min during the ship transits. Bermuda reef  $f\text{CO}_2$  data used in this paper were taken from samples collected over the terrace reef. The  $f\text{CO}_2$  measurements had an accuracy of  $\sim 1 \mu\text{atm}$  (see Bates et al. 1998a for details).  $f\text{CO}_2$  is the fugacity of CO<sub>2</sub> in equilibrium with seawater or atmosphere, which takes account of the nonideal nature of the gas phase. Originally, partial pressure of CO<sub>2</sub> ( $p\text{CO}_2$ ) data was collected and subsequently converted to  $f\text{CO}_2$  values. The difference between  $f\text{CO}_2$  and  $p\text{CO}_2$  is very small (typically  $<0.2\%$ ). For more details on these definitions, see Dickson and Goyet (1994).

The seasonal and interannual variability of CO<sub>2</sub> and the processes controlling the ocean carbon cycle and the air–sea exchange of CO<sub>2</sub> in the Sargasso Sea are well documented (Michaels et al. 1994; Bates et al. 1996a,b, 1998a,b, 2000; Michaels and Knap 1996; Bates 2001; Steinberg et al. 2001). CO<sub>2</sub>, hydrographic, and biogeochemical data have been collected at the BATS site and across the Sargasso Sea since 1988 (<http://www.bbsr.edu>). For this analysis,  $f\text{CO}_2$  data collected at BATS can be considered representative of the Sargasso Sea and thus can be used to constrain our knowledge of how Sargasso Sea waters contacted by the Bermuda coral reef ecosystem are modified.

*Other parameters of the carbonate system*—Supplemental TCO<sub>2</sub> and alkalinity data are also used in the analysis here. TCO<sub>2</sub> was determined by coulometry with a precision of 0.025% ( $\sigma = 0.4 \mu\text{mol kg}^{-1}$ ) based on duplicate and triplicate analyses of  $>1,500$  samples at BBSR from 1991 to 1998 (Bates et al. 1996a). Total alkalinity was determined by potentiometric titration using a modified Gran plot with a precision of 0.2% ( $\sim 4 \mu\text{mol kg}^{-1}$ ) (see Bates et al. 1996a for details).

*Air–sea CO<sub>2</sub> fluxes*—The exchange of CO<sub>2</sub> between ocean and atmosphere is driven by differences in  $f\text{CO}_2$  at the air–sea interface. The net air–sea flux of CO<sub>2</sub> ( $F$ ) is expressed as

$$F = ks(\Delta f\text{CO}_2) \quad (1)$$

where  $k$  is the transfer velocity,  $s$  is the solubility of CO<sub>2</sub>, and  $\Delta f\text{CO}_2$  is the difference between atmosphere and ocean. The variable  $k$  is commonly parameterized as a function of wind speed (e.g., Liss and Merlivat 1986; Wanninkhof 1992; Wanninkhof and McGillis 1999), and CO<sub>2</sub> fluxes can vary by up to a factor of two depending on the  $k$ –wind speed relationship used. For the comparison of CO<sub>2</sub> fluxes on the Bermuda reef and surrounding Sargasso Sea, the quadratic  $k$ –wind speed relationship of Wanninkhof (1992) is used.

$$k = 0.31u^2(660/\text{Sc})^{1/2} \quad (2)$$

$u$  is wind speed at 10 m above mean sea level, and  $\text{Sc}$  is the Schmidt number for CO<sub>2</sub>.  $\text{Sc}$  was calculated using the equations of Wanninkhof (1992), and  $s$  (solubility of CO<sub>2</sub> per unit volume of seawater) was calculated from the observed temperature and salinity using the equations of Weiss (1974). Wind speed data for the period 1994–1996, collected at approximately 10.5 m above the former Naval Air Station on Bermuda, were corrected to 10 m using the equations of Smith (1988).

## Results and discussion

*Seasonal variability of  $f\text{CO}_2$  and temperature*—The Bermuda reef system undergoes considerable seasonal forcing. The seasonal patterns of surface temperature were similar for both the Bermuda reef and the Sargasso Sea (Fig. 2a), although the temperature ranges were different. The seasonal temperature range for the Bermuda reef was  $\sim 12^\circ\text{C}$  compared to  $\sim 8^\circ\text{C}$  at the BATS site. The shallow waters of Bermuda were cooler in winter ( $\sim 17$ – $19^\circ\text{C}$ ) than the Sargasso Sea ( $\sim 19$ – $21^\circ\text{C}$ ). Similar summertime temperature ranges occurred ( $\sim 26$ – $29^\circ\text{C}$ ) at both sites, with Bermuda waters marginally warmer.

Seawater  $p\text{CO}_2$  conditions varied seasonally at both sites (Fig. 2b). The seasonal range on the Bermuda reef was  $\sim 120$ – $150 \mu\text{atm}$ , compared to  $\sim 70$ – $90 \mu\text{atm}$  in the Sargasso Sea. Winter  $p\text{CO}_2$  minima at both sites were similar ( $\sim 310$ – $330 \mu\text{atm}$ ), but summer maxima were markedly higher on the platform ( $\sim 400$ – $460 \mu\text{atm}$ ) compared to the Sargasso Sea ( $\sim 380$ – $410 \mu\text{atm}$ ).

*Modification of seawater  $f\text{CO}_2$  on the Bermuda coral reef*—The Sargasso Sea surrounding Bermuda is the primary source of water mixed by wind and tide onto the Bermuda platform and coral reef system. Any difference in the CO<sub>2</sub> properties between the Bermuda platform and the Sargasso Sea thus reflects the modification of the original source waters by a combination of physical (e.g., temperature) and biological processes (e.g., coral reef CaCO<sub>3</sub> production). Seawater  $f\text{CO}_2$  on the Bermuda reef was typically  $\sim 0$ – $50 \mu\text{atm}$  higher than in the Sargasso Sea (Fig. 2b). Such differences in seawater  $f\text{CO}_2$  were not, however, related to temperature gradients between offshore waters of the Sargasso Sea and reef waters of Bermuda. The effect of warming or cooling on seawater  $f\text{CO}_2$  can be accounted for by correcting  $f\text{CO}_2$  data to a constant temperature using an empirical thermodynamic correction of 4.23% change in  $f\text{CO}_2$  per  $1^\circ\text{C}$  change (Takahashi et al. 1993). This analysis indicates that temperature-corrected seawater  $f\text{CO}_2$  from the Bermuda reef was typically higher than in the Sargasso Sea (Fig. 2c) by as much as  $50 \mu\text{atm}$  (Fig. 2d). Minor day-to-day variability in the modification of  $f\text{CO}_2$  (up to  $\sim 10$ – $20 \mu\text{atm}$ ) on the Bermuda reef was evident (Fig. 2d), which presumably relates to variability in the mixing and modification history of water resident on the reef system during shipboard sampling.

The residence time of water on the eastern part of the Bermuda reef system (i.e., the area sampled in this study) is typically 1–3 tidal cycles (R. J. Johnson unpubl. data). Reef

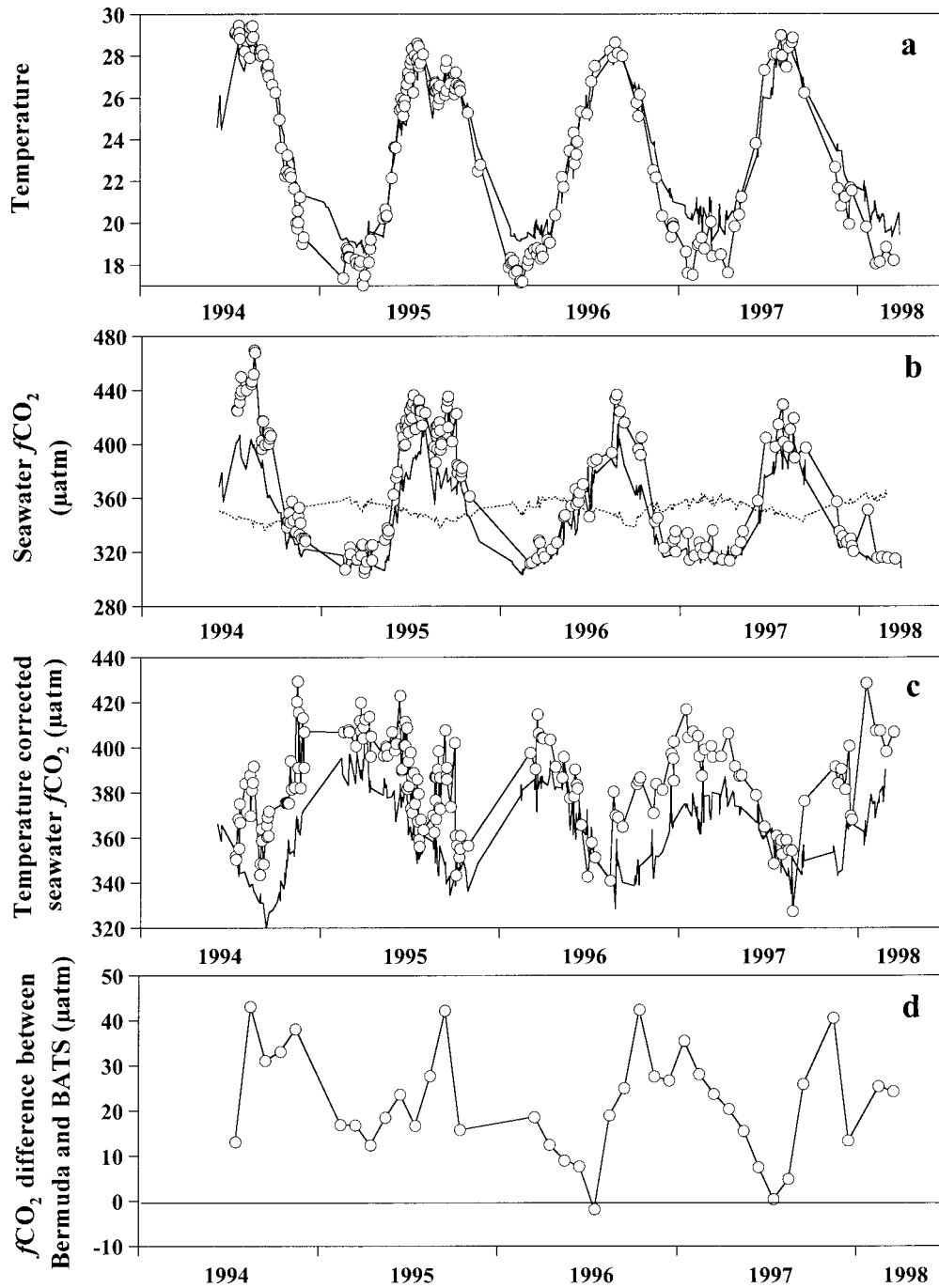


Fig. 2. Time series of temperature and  $\text{CO}_2$  properties in the Bermuda coral reef and Sargasso Sea (i.e., BATS site). (a) Time series of temperature ( $^{\circ}\text{C}$ ) data from the Bermuda coral reef (open circle) and BATS (solid line). (b) Seawater  $f\text{CO}_2$  ( $\mu\text{atm}$ ) data from the Bermuda coral reef (open circle) and BATS (solid line). Atmospheric  $f\text{CO}_2$  data is plotted as dotted line. (c) Seawater  $f\text{CO}_2$  ( $\mu\text{atm}$ ) data from the Bermuda coral reef (open circle) and BATS (solid line) corrected to a constant temperature of  $25^{\circ}\text{C}$  using the thermodynamic equation of 4.23% change in  $p\text{CO}_2$  per  $^{\circ}\text{C}$  change (Takahashi et al. 1993). Each data point for the BATS data is the mean daily temperature and  $f\text{CO}_2$  in the vicinity of BATS and Hydrostation S. Each data point for the Bermuda Platform represents the mean temperature and  $f\text{CO}_2$  of data collected while the R/V *Weatherbird II* was on the Platform. (d) Difference between BATS and Bermuda Platform seawater  $f\text{CO}_2$  corrected to a constant temperature of  $25^{\circ}\text{C}$ . Each data point represents the mean difference calculated for each month over the 4 yr of observation. Seawater was pumped from a bow intake located approximately 1 m deep. Temperature and conductivity were measured continuously using a SeaBird thermosalinograph system with probes placed at the bow intake and just before the pumping system. Seawater and atmospheric  $p\text{CO}_2$  data was measured 12 and 2 times per hour, respectively. Seawater  $p\text{CO}_2$  data was calibrated every 30 min with four  $\text{CO}_2$ -in-dry-air standards referenced against World Meteorological Organization (WMO) standards. A slight warming ( $\sim 0.05^{\circ}\text{C}$ ) was observed between bow intake and equilibrator. The warming correction for  $p\text{CO}_2$  was within the analytical uncertainty of the  $p\text{CO}_2$  measurement ( $\sim 1 \mu\text{atm}$ ; Bates et al. 1998a)

waters in this location are thus continually renewed from offshore sources within a couple of days, requiring that the observed modification of seawater  $f\text{CO}_2$  on the reef occur quickly (i.e., within a few days). In the North Lagoon (see Fig. 1), offshore waters can reside onshore for longer periods ( $\sim 5$ – $10$  d), and the enhancement of reef seawater  $f\text{CO}_2$  relative to the Sargasso Sea can exceed  $100 \mu\text{atm}$  (Bates et al. 2001).

The increase in seawater  $f\text{CO}_2$  on the Bermuda reef system reflects the net balance of several different processes, including (1) air–sea gas exchange of CO<sub>2</sub>, (2) water column net primary production, (3) vertical mixing processes, (4) benthic organic carbon production, and (5) CaCO<sub>3</sub> production. Analysis of the potential contribution of these five processes to changes in seawater  $f\text{CO}_2$  on the reef system indicates that the first three terms (i.e., gas exchange, water column productivity, and vertical mixing) have a negligible effect on CO<sub>2</sub> dynamics. For example, air–sea fluxes of CO<sub>2</sub> in the Sargasso Sea generally range between  $+5$  and  $-5 \text{ mmol CO}_2 \text{ m}^2 \text{ d}^{-1}$ , depending on the seasonal variability of wind and  $\Delta f\text{CO}_2$  (e.g., Bates et al. 1998a). Considering that the residence of water on this part of the reef is short ( $< 2$  d), such rates of CO<sub>2</sub> flux will change seawater  $f\text{CO}_2$  by only a small amount ( $< \sim 1$ – $2 \mu\text{atm d}^{-1}$ ). The rates of water column primary production on the Bermuda coral reef are low (Morris et al. 1977), slightly higher than rates in the surrounding Sargasso Sea (Michaels and Knap 1996; Steinberg et al. 2001). Although rates of <sup>14</sup>C primary production were not measured on the reef during the period of CO<sub>2</sub> sampling, chlorophyll *a* (Chl *a*) biomass, an indirect proxy for production, was measured in North Lagoon from 1996 to 1998 (Connelly 1997; D. P. Connelly unpubl.). Surface Chl *a* biomass seasonally ranged from  $0.1$  to  $0.4 \text{ mg C m}^{-3}$  in North Lagoon compared to chlorophyll biomass of  $0.02$ – $0.4 \text{ mg C m}^{-3}$  observed at BATS (Steinberg et al. 2001). In the Sargasso Sea, diurnal variability of seawater TCO<sub>2</sub> and  $f\text{CO}_2$  due to plankton production and respiration at BATS is minor ( $\sim 0.3 \mu\text{mol kg}^{-1}$  TCO<sub>2</sub> and  $< 1 \mu\text{atm } f\text{CO}_2$ ; Bates et al. 1996, 1998; N. R. Bates unpubl. data). Similarly, diurnal variability of seawater  $f\text{CO}_2$  on the Bermuda reef system should also be minor ( $\sim 1$ – $2 \mu\text{atm}$ ) given the low rates of water column production on the Bermuda reef system. In addition, the contribution of water column CaCO<sub>3</sub>-producing phytoplankton, such as *Emiliania huxleyi*, to seawater CO<sub>2</sub> dynamics should be negligible considering the minor contribution of calcifying plankton to pelagic plankton found on the Bermuda reef system (Morris et al. 1977). Finally, vertical mixing processes will not contribute significantly to the observed seawater CO<sub>2</sub> changes because the reef waters are shallow ( $< 8.5$  m) and there is a lack of vertical thermohaline or CO<sub>2</sub> gradient onshore.

The modification of seawater  $f\text{CO}_2$  (i.e., up to  $50 \mu\text{atm}$ ) on the Bermuda coral reef largely reflects the net balance between organic carbon production (decreasing  $f\text{CO}_2$ ) and CaCO<sub>3</sub> production (increasing  $f\text{CO}_2$ ). Calcification in the daylight has been measured at a rate of  $12.3 \pm 3.6 \text{ mmol m}^{-2} \text{ h}^{-1}$  in October 1998 (Bates et al. 2001) at Hog Reef Flat, which is approximately  $10$ – $12$  km northwest of the R/V *Weatherbird II* sampling route on the Bermuda reef. This rate of calcification is comparable to rates measured ( $\sim 6$ –

$16 \text{ mmol m}^{-2} \text{ h}^{-1}$ ) for *Diploria* from rim reef sites by Dodge and coworkers using buoyant weight techniques (Dodge et al. 1984, 1985). Their calcification rates ( $\text{g d}^{-1}$ ) were converted ( $\text{mmol m}^{-2} \text{ h}^{-1}$ ) assuming a hard coral coverage of  $20$ – $30\%$ . Calcification rates measured on other reef systems have similar ranges (e.g., Kinsey 1985; Gattuso et al. 1993b; 1997).

Seawater changes in TCO<sub>2</sub> and TA concentrations are also useful as qualitative indicators of calcification because  $\Delta\text{TCO}_2:\Delta\text{TA}$  ratios (normalized to a constant salinity) should theoretically range between  $1:1$  and  $1:2$  (variable depending on the balance of organic carbon to CaCO<sub>3</sub> production). In this analysis, it is assumed that the low rates of water column production and respiration do not contribute to  $\Delta\text{TCO}_2$  changes (see above discussion). A  $\Delta\text{TCO}_2:\Delta\text{TA}$  change of  $\sim 1:1.13$  was observed during a diurnal time series at Hog Reef Flat, part of the northwest rim reef (Bates et al. 2001). Seawater  $f\text{CO}_2$  and temperature-corrected  $f\text{CO}_2$  data measured using an autonomous  $f\text{CO}_2$  buoy system were also higher by  $20$ – $60 \mu\text{atm}$  at this site compared to the Sargasso Sea (Bates et al. 2001). Along the ship track, a decrease in TCO<sub>2</sub> and TA of  $5$  and  $12 \mu\text{mol kg}^{-1}$ , respectively, between offshore and onshore was observed in March 1996. The  $\Delta\text{TCO}_2:\Delta\text{TA}$  ratio of  $\sim 1:2$  suggests that the influence of CaCO<sub>3</sub> production on the CO<sub>2</sub> system was greater than organic carbon production at this time.

Analyses of organic and CaCO<sub>2</sub> composition of reef sediments can also provide long-term evidence about the dynamics of CO<sub>2</sub> on coral reefs. Reef systems with sediments having a weight percentage of organic carbon greater than  $6\%$  should theoretically be net sinks for CO<sub>2</sub>, whereas reefs with sediments having a low organic carbon content ( $< 6\%$ ) should be net sources of CO<sub>2</sub> to the atmosphere (Buddemeier 1996). Sediments from a variety of reef settings on Bermuda have a variable but low organic carbon content of  $\sim 3$ – $7\%$  (Jickells and Knap 1984), indicating that balance of processes on the Bermuda reef system releases CO<sub>2</sub> to the waters overlying the reef.

*Seasonal patterns of CO<sub>2</sub> dynamics, calcification, and productivity on the Bermuda coral reef*—The Bermuda reef is subject to seasonal change (i.e., temperature, irradiance, and community structure), and there are seasonal changes in the release of CO<sub>2</sub> from the reef to overlying waters (Fig. 3a). These changes reflect changes in the balance of organic carbon and CaCO<sub>3</sub> production that reflect seasonal changes in the reef community structure (i.e., balance of calcifying hard corals and organic carbon-producing macroalga). The modification of CO<sub>2</sub> on Bermuda reef waters reaches a broad maximum during the fall–winter period when Bermuda reef seawater  $f\text{CO}_2$  was  $+30$ – $+50 \mu\text{atm}$  higher than offshore conditions. The smallest or zero net modification of Bermuda reef  $f\text{CO}_2$  occurs in the May–July period (see Fig. 3a), coincident with the period (i.e., June) of maximum solar input to the sea surface. During this period, the processes of organic carbon and CaCO<sub>3</sub> production compensate each other with almost no uptake or release of CO<sub>2</sub> to the water. Occasionally, the reef community becomes a sink for CO<sub>2</sub> (i.e., July 1996 and 1997; see Fig. 3a).

Rates of calcification and growth by extension for several

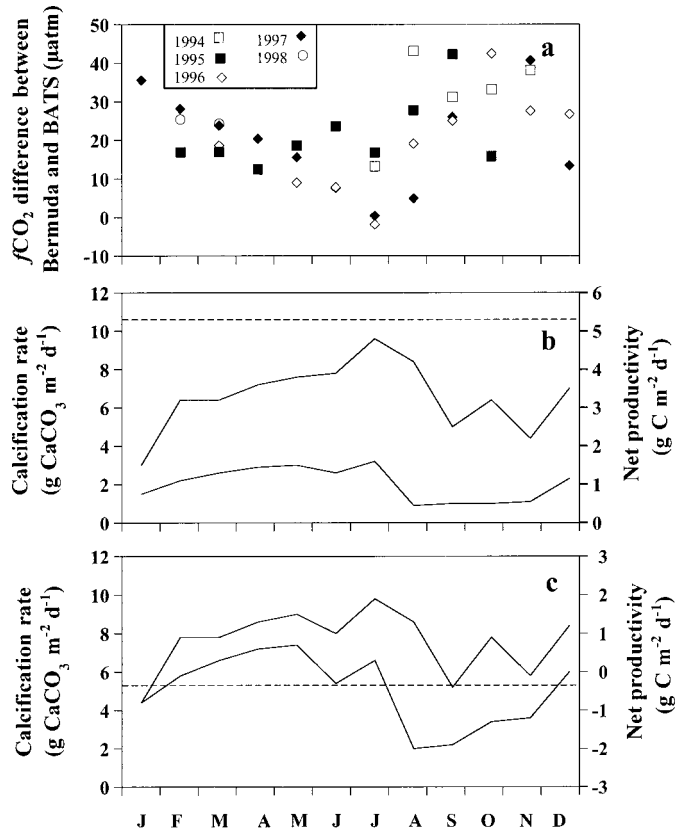


Fig. 3. Seasonal patterns of  $\text{CO}_2$  differences, calcification, and net productivity. (a) Mean monthly difference between BATS and Bermuda Platform seawater  $f\text{CO}_2$  ( $\mu\text{atm}$ ) plotted against decimal year. The seawater  $f\text{CO}_2$  data is corrected to a constant temperature of  $25^\circ\text{C}$ . Positive numbers indicate that Bermuda Platform seawater  $f\text{CO}_2$  was higher than source waters at BATS. (b) High and low rates of net productivity (solid lines,  $\text{g C m}^{-2} \text{ d}^{-1}$ ) computed from range of mean monthly seawater  $f\text{CO}_2$  for the period 1994–1998, and a constant high calcification rate of  $10.6 \text{ g CaCO}_3 \text{ m}^{-2} \text{ d}^{-1}$  (dashed line). (c) High and low rates of net productivity (solid lines,  $\text{g C m}^{-2} \text{ d}^{-1}$ ) computed from range of mean monthly seawater  $f\text{CO}_2$  for the period 1994–1998, and a constant low calcification rate of  $5.3 \text{ g CaCO}_3 \text{ m}^{-2} \text{ d}^{-1}$  (dashed line).

hard coral species on Bermuda have been measured directly (e.g., Dodge et al. 1984, 1985; Logan and Tomascik 1991; Logan et al. 1994), whereas rates of macroalgal productivity (as an index for rate of organic carbon production) have not. Given knowledge of calcification and the seasonal dynamics of  $\text{CO}_2$ , rates of macroalgal productivity and the seasonal patterns of productivity on the reef can be quantified using carbon mass balance considerations (shown below).

Coral growth and extension rates on the Bermuda reef show distinct spatial variability and seasonality. Several studies have observed that annual coral extension rates were higher in the inshore and nearshore patch reefs of North Lagoon, compared to rim reef at the outer edge of the Bermuda Platform (Dodge and Vaisnys 1977; Logan and Tomascik 1991; Logan et al. 1994). Seasonally, two of the major framework-building coral species, *D. labyrinthiformis* and *Porites astreoides*, located in the outer rim reef, accrete narrow, high-density bands during the summer and wider

low-density bands during the fall to spring (Logan and Tomascik 1991; A. L. Cohen et al. pers. comm.).

Calcification has been determined for Bermuda corals such as *D. labyrinthiformis* at a rate of  $0.2\text{--}0.4 \text{ g CaCO}_3 \text{ d}^{-1}$  for 12-cm-diameter (size-normalized) specimens (Dodge et al. 1984, 1985). In these studies, no statistical difference between winter and summer calcification was discerned. Furthermore, although extension rates differ seasonally (Logan and Tomascik 1991; Logan et al. 1994), the mass of  $\text{CaCO}_3$  accreted per unit time appears similar in both the winter wide, low-density band and summer narrow, high-density band (A. L. Cohen et al. pers. comm.). Consistent with the above studies, it is assumed that the calcification rate of hard corals in Bermuda remains fairly constant during the year. The rates of calcification measured by Dodge and coworkers can be extrapolated to an areal basis. In the southeastern outer rim, hard corals have  $\sim 30\%$  cover (S. R. Smith unpubl. data), slightly higher than the mean hard coral cover on Bermuda (CARICOMP 1997). The rates of calcification are estimated at  $5.6$  and  $10.6 \text{ g CaCO}_3 \text{ m}^{-2} \text{ d}^{-1}$ , values consistent with the measured range of  $\text{CaCO}_3$  production (Dodge et al. 1984, 1985). Calcification rates on other reefs tend to be higher ( $5\text{--}40 \text{ g CaCO}_3 \text{ m}^{-2} \text{ d}^{-1}$  (e.g., Barnes and Lazar 1993; Gattuso et al. 1993b, 1996b, 1999b), but the extension rates of Bermuda corals are lower ( $\sim 3 \text{ mm yr}^{-1}$ ; Logan and Tomascik 1991; Logan et al. 1994) compared to corals at low-latitude reefs ( $\sim 5\text{--}15 \text{ mm yr}^{-1}$ ).

As discussed earlier, organic carbon and  $\text{CaCO}_3$  production are the two primary determinants of  $\text{CO}_2$  dynamics on the Bermuda reef. In a mass balance sense, if there is no net gain or loss of  $\text{CO}_2$  in waters resident on the reef, the effect of organic carbon production and  $\text{CaCO}_3$  production on seawater  $f\text{CO}_2$  is in balance and equal. That is,

$$\text{if } \Delta f\text{CO}_{2\text{calc}} = \Delta f\text{CO}_{2\text{prod}}, \text{ then } \Delta f\text{CO}_{2\text{obs}} = 0 \quad (3)$$

where  $\Delta f\text{CO}_{2\text{obs}}$  is the observed change in seawater  $f\text{CO}_2$  and  $\Delta f\text{CO}_{2\text{calc}}$  and  $\Delta f\text{CO}_{2\text{prod}}$  are the effect of calcification and organic carbon production, respectively, on seawater  $f\text{CO}_2$ . Given this condition, net productivity can be estimated at  $2.3$  and  $4.6 \text{ g C m}^{-2} \text{ d}^{-1}$  (estimated from the two rates on calcification). Here, net productivity is primarily attributable to macroalgal production. Because of the short residence time (less than 2 d) of water in this part of the Bermuda reef, the contribution of other factors, such as water column productivity and air–sea  $\text{CO}_2$  gas exchange, were minor contributors to the productivity estimates ( $<5\%$ ).

For most of the year, the effect of organic carbon production and  $\text{CaCO}_3$  production on  $\text{CO}_2$  thermodynamics is not in balance, and reef metabolism causes a considerable gain of  $\text{CO}_2$  compared to the Sargasso Sea (Fig. 3a).

$$\text{If } \Delta f\text{CO}_{2\text{calc}} > \Delta f\text{CO}_{2\text{prod}}, \text{ then } \Delta f\text{CO}_{2\text{obs}} > 0 \quad (4)$$

$$\Delta f\text{CO}_{2\text{prod}} = \Delta f\text{CO}_{2\text{calc}} - \Delta f\text{CO}_{2\text{obs}} \quad (5)$$

Rates of net productivity can thus be determined from  $\Delta f\text{CO}_{2\text{calc}}$  and  $\Delta f\text{CO}_{2\text{obs}}$  with the prerequisite that carbon mass balance is maintained. Given the two bounding rates of calcification ( $5.6$  and  $10.6 \text{ g CaCO}_3 \text{ m}^{-2} \text{ d}^{-1}$ ), seasonal rates of net productivity can be determined from monthly  $\Delta f\text{CO}_{2\text{obs}}$  data and knowledge of  $\text{CO}_2$  thermodynamics (Fig.

3b,c). The lowest and highest seawater  $f\text{CO}_2$  changes (Fig. 3a) each month (for the 1994–1998 period) are used to bound the range of  $\Delta f\text{CO}_{2\text{obs}}$  and rates of net productivity. It is also assumed that water on the reef is renewed every 2 d. These two simplifications smooth out the day-to-day variability in net productivity (due to changing light conditions, residence times, etc.), as evidenced in variability of  $\Delta f\text{CO}_{2\text{obs}}$  (Fig. 2c).

This carbon mass balance analysis reveals seasonal patterns of macroalgal net productivity (Fig. 3b,c). During summertime, macroalgal net productivity reaches a maximum rate ( $\sim 1\text{--}4.6 \text{ g C m}^{-2} \text{ d}^{-1}$ ) and the system is net autotrophic. This seasonal maximum is coincident with the June solstice period of highest potential irradiance and a maximum in macroalgal biomass (S. R. Smith unpubl. data). Other studies have similarly shown that macroalgal productivity increases in relation to solar irradiance and seawater temperature (e.g., Gattuso et al. 1997). In the macroalga-dominated reef communities of Moorea, organic carbon production exceeded CaCO<sub>3</sub> production (and the reef acted as a sink for CO<sub>2</sub>) during periods of highest irradiance (Gattuso et al. 1997). On the Bermuda reef, organic carbon production occasionally exceeded (in June 1995 and 1996) CaCO<sub>3</sub> production (or  $\Delta f\text{CO}_{2\text{prod}} \geq \Delta f\text{CO}_{2\text{calc}}$ ) and the reef briefly switched to a sink for CO<sub>2</sub>.

After summer each year, macroalgal biomass on the Bermuda reef declines (as solar input decreases and percent cloud cover increases) (S. V. Smith unpubl. data). A seasonal minimum in net productivity occurs during winter and spring (Fig. 3b,c), coincident with the seasonal minimum of macroalgal biomass (S. R. Smith unpubl. data). After summer, reef metabolism releases CO<sub>2</sub>, with a maximum difference in seawater  $f\text{CO}_2$  reached by October (Fig. 3a). Although most CO<sub>2</sub> is released because of calcification, negative rates of net productivity during the fall–spring period (Fig. 3b,c) suggest (if the low rates of calcification are correct) that the system becomes heterotrophic, with organic carbon respiration contributing CO<sub>2</sub> to reef waters.

The seasonal variability of CO<sub>2</sub> processing on the Bermuda reef is due to changes in the balance of organic carbon and CaCO<sub>3</sub> production, which in turn appears to reflect seasonal variability in the balance of the reef ecosystem (i.e., between hard corals and macroalgae). On an annual basis, reef calcification overwhelms organic carbon production, and CO<sub>2</sub> is released to the water column. The dynamics of CO<sub>2</sub> and carbon mass balance considerations thus provide an overall, integrated, seasonal view of reef metabolism on Bermuda.

Because of assumptions and uncertainties of mass balance, the absolute rates of net macroalgal productivity remain uncertain and require comparison to direct measurements. The net productivity rate estimates for the Bermuda reef system were similar to rates on other reefs. Given the two bounding rates of calcification (5.6 and 10.6 g CaCO<sub>3</sub> m<sup>-2</sup> d<sup>-1</sup>) and the monthly differences in  $\Delta f\text{CO}_{2\text{obs}}$ , mean net productivity ranges from  $-0.6$  to  $3.3 \text{ g C m}^{-2} \text{ d}^{-1}$  (Table 1). The reef system appears to seasonally shift between autotrophy and heterotrophy, with an overall tendency toward annual net autotrophy. Net and gross productivity measurements on other reefs range from 0.0 to  $1.5 \text{ g C m}^{-2} \text{ d}^{-1}$  (e.g.,

Table 1. Mean calcification (g CaCO<sub>3</sub> m<sup>-2</sup> d<sup>-1</sup>) and net productivity (g C m<sup>-2</sup> d<sup>-1</sup>) rates averaged on an annual basis.

	High calcification	Low calcification
Calcification rate (g CaCO <sub>3</sub> m <sup>-2</sup> d <sup>-1</sup> )	10.6	5.3
Net productivity* (g C m <sup>-2</sup> d <sup>-1</sup> )	2.0–3.3	–0.5–1.0
Net productivity† (g C m <sup>-2</sup> d <sup>-1</sup> )	1.3–2.5	–0.6–0.6

\* Net productivity is calculated assuming that the rate of calcification is constant throughout the year.

† Net productivity is calculated assuming that the rate of calcification is lower during the summer months. The calcification rate for October to May is assumed as above (i.e., 10.6 and 5.3 g CaCO<sub>3</sub> m<sup>-2</sup> d<sup>-1</sup>, respectively). For the period June–September, the calcification rate is halved (i.e., 5.3 and 2.6 g CaCO<sub>3</sub> m<sup>-2</sup> d<sup>-1</sup>, respectively).

Kinsey 1985; Chisholm et al. 1990; Crossland et al. 1991) and 2.3 to  $>10 \text{ g C m}^{-2} \text{ d}^{-1}$  (Kinsey 1985; Crossland et al. 1991; Barnes and Lazar 1993; Gattuso et al. 1993b, 1996b, 1999b, 2000), respectively. If summertime calcification rates of Bermuda coral are lower during accretion of the narrow but higher density band, net productivity rates will be closer to 0 ( $-0.6$  to  $2.5 \text{ g C m}^{-2} \text{ d}^{-1}$ ; Table 1).

What remains uncertain are the seasonal rates and spatial variability (e.g., outer rim reef to inshore/nearshore patch reef) of productivity and calcification (i.e., organic carbon and CaCO<sub>3</sub> production) for individual hard coral and macroalgal species components of the collective ecosystem. However, the “integrated view” of seasonal CO<sub>2</sub> dynamics in response to Bermuda reef metabolism may be useful for understanding carbon cycling in other reefs, particularly those that appear to be in transition from coral-dominated communities to macroalga-dominated communities (e.g., Kinsey 1985; Smith and Buddemeier 1992; Hughes 1994).

There is some evidence that Bermuda reef metabolism may also be responding to interannual variability. In June 1996 and 1997 (compared to 1994 and 1995), reef metabolism shifted to a slight uptake of CO<sub>2</sub>, indicative of a shift in the balance of organic carbon and CaCO<sub>3</sub> production (Fig. 2d). Such changes may be related to interannual changes in the subtropical gyre surrounding the island of Bermuda and in the supply of nutrients to the reef system. Interannual variability in temperature, salinity, primary production, mixed-layer depths, and vertical nitrate supply has been demonstrated (Bates 2001), and much of the interannual variability relates to modes of climate variability such as El Niño–Southern Oscillation (ENSO) and North Atlantic Oscillation (NAO). For example, in 1996, in response to a shift in NAO phase to a negative state in the North Atlantic, the Sargasso Sea was cooler (by  $\sim 0.5^\circ\text{C}$ ), mixed layers were deeper (up to 30–40 m), vertical mixing of nitrate was greater, and rates of primary production were higher (up to  $200 \text{ mg C m}^{-2} \text{ d}^{-1}$ ) compared to the mean state (Bates 2001). During this period, the oceanic uptake of CO<sub>2</sub> in the Sargasso Sea appears to have increased (N. Gruber et al. pers. comm.). Contemporaneous and consistent with this temporal variation of air–sea CO<sub>2</sub> exchange, the loss of CO<sub>2</sub> from the Bermuda reef system also seems to have been reduced. With longer-term reef records, mechanistic relationships between

Table 2. Estimates of air–sea gas exchange of CO<sub>2</sub> on the Bermuda Platform and in the Sargasso Sea at BATS. Air–sea CO<sub>2</sub> fluxes were calculated using the wind speed–CO<sub>2</sub> transfer velocity relationships of Wanninkhof (1992). Wind speed data was collected at a height of 10.5 m at the former U.S. Naval Air Station, St. David's, Bermuda.

Sampling period	Location	CO <sub>2</sub> flux* (mmol CO <sub>2</sub> m <sup>-2</sup> yr <sup>-1</sup> )	Mean Δ <i>f</i> CO <sub>2</sub> † (μatm)
10 Jul 1994–31 Dec 1994‡	Platform	+0.13	+32.9
	BATS	-0.13	+6.9
1 Jan 1995–31 Dec 1995§	Platform	-0.12	+7.4
	BATS	-0.73	-8.8
1 Jan 1996–31 Dec 1996	Platform	-0.11	+5.9
	BATS	-0.85	-10.1

\*Negative values denote CO<sub>2</sub> flux from atmosphere to ocean; positive values denote flux from ocean to atmosphere.

† Δ*f*CO<sub>2</sub> is difference between ocean and atmospheric *f*CO<sub>2</sub>, where negative values denote *f*CO<sub>2</sub> values higher in the atmosphere and positive values denote *f*CO<sub>2</sub> values higher in the ocean.

‡ Although Δ*f*CO<sub>2</sub> was on average positive in 1994 at BATS, the flux was negative and directed into the ocean because of higher CO<sub>2</sub> flux in the fall when Δ*f*CO<sub>2</sub> was negative but wind speeds were much greater compared to the summer.

§ In 1995 and 1996, the CO<sub>2</sub> flux and Δ*f*CO<sub>2</sub> values were also opposite in sign for the Bermuda platform. In both years, Δ*f*CO<sub>2</sub> was on average positive, but the flux was negative (again because of higher wind speeds in the winter and spring when Δ*f*CO<sub>2</sub> was negative).

|| Wind speed data was not available after 1996.

calcification, macroalgal productivity, and climate forcing may be clearly demonstrated in the future.

*The Bermuda coral reef: Source or sink of CO<sub>2</sub> to the atmosphere?*—There has been considerable debate about whether coral reefs are sources to the atmosphere (e.g., Ware et al. 1992; Gattuso et al. 1993a, 1995, 1996a, 1999b) or oceanic sinks of CO<sub>2</sub> (e.g., Yamamuro et al. 1995; Kayanne et al. 1996; Gattuso et al. 1997; Kraines et al. 1997). Most studies, with notable exceptions (Gattuso et al. 1995, 1996b, 1999b; Frankignoulle et al. 1996), have assumed that reef metabolism (through release or uptake of CO<sub>2</sub>) is the only factor influencing air–sea CO<sub>2</sub> flux and the sink or source status of reefs. A contributing factor to the source or sink status of reefs, however, is the pre-existing air–sea CO<sub>2</sub> disequilibrium of offshore waters impinging on the reef system.

The Bermuda reef ecosystem, particularly the outer rim and terrace reefs, is predominantly a source of CO<sub>2</sub> to the water overlying the reef, but not necessarily to the atmosphere. Because of the short residence time of water on the outer rim and terrace reef, reef metabolism modifies the pre-existing air–sea CO<sub>2</sub> disequilibrium of waters impinging the reef. The surrounding Sargasso Sea undergoes regular seasonal changes in seawater *f*CO<sub>2</sub> and the direction of air–sea CO<sub>2</sub> exchange (Bates et al. 1998a; Bates 2001). In winter, Sargasso Sea seawater *f*CO<sub>2</sub> is lower by ~40–60 μatm than the atmosphere, and CO<sub>2</sub> exchange is directed from atmosphere to ocean (i.e., a net oceanic sink of CO<sub>2</sub>). The release of CO<sub>2</sub> from Bermuda reef calcification elevates seawater *f*CO<sub>2</sub> on waters impinging the reef, reducing the rate of atmosphere to ocean CO<sub>2</sub> flux (Table 2), but not reversing

the direction of gas exchange. In summer, the air–sea CO<sub>2</sub> disequilibrium of offshore water is reversed (i.e., seawater *f*CO<sub>2</sub> is higher than atmosphere *f*CO<sub>2</sub> by 40–60 μatm), and the release of CO<sub>2</sub> from reef metabolism augments the rate of sea to air CO<sub>2</sub> flux from the waters of Bermuda (Table 2).

The influence of the pre-existing air–sea CO<sub>2</sub> disequilibrium of offshore water masses impinging on the reef has also been demonstrated on the Moorea Reef in the Pacific Ocean (Gattuso et al. 1995, 1996b; Frankignoulle et al. 1996). Two visits in winter and summer to this reef system showed considerable differences in offshore *f*CO<sub>2</sub> conditions presumably reflecting seasonal changes in air–sea CO<sub>2</sub> disequilibrium. Regional maps of the air–sea CO<sub>2</sub> disequilibrium in the tropical/subtropical Pacific (Takahashi et al. 1997) also reveal large seasonal variation (up to 60–80 μatm) of seawater *p*CO<sub>2</sub> in offshore waters surrounding Pacific Ocean coral reef sites. Thus, factors such as seasonal offshore air–sea CO<sub>2</sub> disequilibrium, mixing between coral reef and offshore waters and the seasonality of biological processes (i.e., net balance of calcification and organic carbon production) must be taken into account when evaluating the regional and global significance of coral reefs to the transfer of CO<sub>2</sub> between the ocean and atmosphere.

CO<sub>2</sub> data from the Bermuda reef provide broad hints about the complexity of the processes that affect CO<sub>2</sub> on coral reef ecosystems. Although the geographic extent of Bermuda coral reefs are small and its coral diversity arguably is limited, the Bermuda reef (and other reefs) are subject to seasonal forcing and changes in community structure that determine the fate and dynamics of CO<sub>2</sub>. The Bermuda reef system serves as a useful model for understanding changes in carbon cycling in other reefs that are in transition from coral-dominated to macroalga-dominated communities in response to environmental stress. A detailed understanding of the complex physical and biological interactions affecting CO<sub>2</sub> occurring on different types of reef ecosystems is warranted.

## References

- BARNES, D. J., AND B. LAZAR. 1993. Metabolic performance of a shallow reef patch near Eilat on the Red Sea. *J. Exp. Mar. Biol. Ecol.* **174**: 1–13.
- BATES, N. R. 2001. Interannual changes of oceanic CO<sub>2</sub> and biogeochemical properties in the Western North Atlantic subtropical gyre. *Deep-Sea Res. II* **48**: 1507–1528.
- , L. SAMUELS, AND L. MERLIVAT. 2001. Biogeochemical and physical factors influencing seawater *f*CO<sub>2</sub> and air–sea CO<sub>2</sub> exchange on the Bermuda coral reef. *Limnol. Oceanogr.* **46**: 833–846.
- , A. F. MICHAELS, AND A. H. KNAP. 1996a. Seasonal and interannual variability of the oceanic carbon dioxide system at the U.S. JGOFS Bermuda Atlantic Time-series Site. *Deep-Sea Res. II* **43**: 347–383.
- , ———, AND ———. 1996b. Alkalinity changes in the Sargasso Sea: Geochemical evidence of calcification? *Mar. Chem.* **51**: 347–358.
- , T. TAKAHASHI, D. W. CHIPMAN, AND A. H. KNAP. 1998a. Variability of *p*CO<sub>2</sub> on diel to seasonal timescales in the Sargasso Sea. *J. Geophys. Res.* **103**: 15,567–15,585.
- , A. H. KNAP, AND A. F. MICHAELS. 1998b. The effect of

- hurricanes on the local to global air-sea exchange of CO<sub>2</sub>. *Nature* **395**: 58–61.
- , L. MERLIVAT, L. BEAUMONT, AND A. C. PEQUIGNET. 2000. Intercomparison of shipboard and moored buoy fCO<sub>2</sub> measurements: Ground-truthing of the CARIOCA sensor in the Sargasso Sea. *Mar. Chem.* **72**: 239–255.
- BUDEMEIER, R. W. 1996. Coral reefs and carbon dioxide. *Science* **271**: 1298–1299.
- CARICOMP. 1997. CARICOMP monitoring of coral reefs. *Proc. 8th Int. Coral Reef Symp.* **1**: 651–656.
- CHISHOLM, J. R. M., J.-C. COLLINGWOOD, AND E. F. GILL. 1990. A novel in situ respirometer for investigating photosynthesis and calcification in crustose coralline algae. *J. Exp. Mar. Biol. Ecol.* **141**: 15–29.
- CONNELLY, D. P. 1997. Occurrence and behaviour of trace metals in coastal waters of Bermuda, and chromium in the Sargasso Sea. Ph.D. thesis, University of Southampton.
- CROSSLAND, C. J., B. G. HATCHER, AND S. V. SMITH. 1991. Role of coral reefs in global ocean production. *Coral Reefs* **10**: 55–64.
- DICKSON, A. G., AND C. GOYET. [eds.] 1994. Handbook of methods for the analysis of the various parameters of the carbon dioxide system in seawater, version 2. U.S. Dept. of Energy CO<sub>2</sub> Science Team Report, ORNL/CDIAC-74.
- DODGE, R. E., AND J. R. VAISNYS. 1977. Coral populations and growth patterns: Responses to sedimentation and turbidity associated with dredging. *J. Mar. Res.* **35**: 715–730.
- , S. C. WYERS, S. C. FRITH, A. H. KNAP, S. R. SMITH, C. B. COOK, AND T. D. SLEETER. 1984. Coral calcification rates by the buoyant weight technique: Effects of alizarin staining. *J. Exp. Mar. Biol. Ecol.* **75**: 217–232.
- , A. H. KNAP, S. C. WYERS, H. R. FRITH, T. D. SLEETER, AND S. R. SMITH. 1985. The effect of dispersed oil on the calcification rate of the reef-building coral *Diploria strigosa*. *Proc. Fifth Int. Coral Reef Congr.* **6**: 453–457.
- FRANKIGNOULLE, M., J.-P. GATTUSO, R. BIONDO, I. BOURGE, G. COPIN-MONTEGUT, AND M. PICHON. 1996. Carbon fluxes in coral reefs. 2. Eulerian study of inorganic carbon dynamics and measurement of air-sea CO<sub>2</sub> exchanges. *Mar. Ecol. Prog. Ser.* **145**: 123–132.
- GATTUSO, J.-P., D. YELLOWLEES, AND M. LESSER. 1993a. Depth- and light-dependent variation of carbon partitioning and utilization in the zooanthellate scleractinian coral *Stylophora pistillata*. *Mar. Ecol. Prog. Ser.* **92**: 267–276.
- , M. PICHON, B. DELESALLE, AND M. FRANKIGNOULLE. 1993b. Community metabolism and air-sea CO<sub>2</sub> fluxes in coral reef ecosystem (Moorea, French Polynesia). *Mar. Ecol. Prog. Ser.* **96**: 259–267.
- , ———, AND M. FRANKIGNOULLE. 1995. Biological control of air-sea CO<sub>2</sub> fluxes: Effect of photosynthetic and calcifying marine organisms and ecosystems. *Mar. Ecol. Prog. Ser.* **129**: 307–312.
- , M. FRANKIGNOULLE, S. V. SMITH, J. R. WARE, AND R. WOLLAST. 1996a. Coral reefs and carbon dioxide. *Science* **271**: 1298.
- , M. PICHON, B. DELESALLE, C. CANON, AND M. FRANKIGNOULLE. 1996b. Carbon fluxes in coral reefs. I. Lagrangian measurement of community metabolism and resulting air-sea CO<sub>2</sub> disequilibrium. *Mar. Ecol. Prog. Ser.* **145**: 109–121.
- , C. E. PAYRI, M. PICHON, B. DELESALLE, AND M. FRANKIGNOULLE. 1997. Primary production, calcification, and air-sea CO<sub>2</sub> fluxes of a macroalgal-dominated coral reef community (Moorea, French Polynesia). *J. Phycol.* **33**: 729–738.
- , D. ALLEMAND, AND M. FRANKIGNOULLE. 1999a. Photosynthesis and calcification at cellular, organismal and community levels in coral reefs: A review on interactions and control by carbonate chemistry. *Am. Zool.* **39**: 160–183.
- , M. FRANKIGNOULLE, AND S. V. SMITH. 1999b. Measurement of community metabolism and significance in the coral reef CO<sub>2</sub> source-sink debate. *Proc. Nat. Acad. Sci. USA* **96**: 13,017–13,022.
- , S. REYNAUD-VAGANAY, P. FURLA, S. ROMAINE-LIUD, J. JAUBERT, I. BOURGE, AND M. FRANKIGNOULLE. 2000. Calcification does not stimulate photosynthesis in the zooxanthellate scleractinian coral *Stylophora pistillata*. *Limnol. Oceanogr.* **45**: 246–250.
- HUGHES, T. P. 1994. Catastrophes, phase shifts, and large-scale degradation of a Caribbean coral reef. *Science* **271**: 1298–1299.
- JICKELLS, T.D., AND A. H. KNAP. 1984. Trace metal distributions on the Bermuda platform. *Estuar. Coast. Shelf Sci.* **18**: 245–254.
- KAWAHATA, H., A. SUZUKI, AND K. GOTO. 1997. Coral reef ecosystems as a source of atmospheric CO<sub>2</sub>: Evidence from pCO<sub>2</sub> measurements of surface waters. *Coral Reefs* **16**: 261–266.
- KAYANNE, H., A. SUZUKI, AND H. SAITO. 1995. Diurnal changes in the partial pressure of carbon dioxide in coral reef water. *Science* **269**: 214–216.
- , ———, AND ———. 1996. Coral reefs and carbon dioxide. *Science* **271**: 1299–1300.
- KINSEY, D. W. 1985. Metabolism, calcification and carbon production. I. System level studies. *Proc. 5th Int. Coral Reef Congr.* **6**: 505–526.
- KLEYPAS, J. A., R. W. BUDEMEIER, D. ARCHER, J.-P. GATTUSO, C. LANGDON, AND B. N. OPDYKE. 1999. Geochemical consequences of increased atmospheric carbon dioxide on coral reefs. *Science* **284**: 118–120.
- KRAINES, S., Y. SUZUKI, K. YAMADA, AND H. KOMIYAMA. 1997. Separating biological and physical changes in dissolved oxygen concentration in a coral reef. *Limnol. Oceanogr.* **41**: 1790–1799.
- LANGDON, C., AND OTHERS. 2000. Effect of calcium carbonate saturation state on the calcification rate of an experimental coral reef. *Glob. Biogeochem. Cycles* **14**: 639–654.
- LISS, P. S., AND L. MERLIVAT. 1986. Air-sea gas exchange rates: Introduction and synthesis. In P. Buat-Menard [ed.], The role of air-sea exchange in geochemical cycling. NATO Adv. Sci. Inst., C **185**: 113–128.
- LOGAN, A., AND T. TOMASCIK. 1991. Extension growth rates in two coral species from high latitude reefs of Bermuda. *Coral Reefs* **10**: 155–160.
- , L. YANG, AND T. TOMASCIK. 1994. Linear skeletal extension rates in two species of *Diploria* from high-latitude reefs in Bermuda. *Coral Reefs* **13**: 225–230.
- MICHAELS, A.F., AND A. H. KNAP. 1996. Overview of the U.S. JGOFS Bermuda Atlantic Time-series Study and the Hydrostation S program. *Deep-Sea Res. II* **43**: 157–198.
- , N. R. BATES, K. O. BUESSELER, C. A. CARLSON, AND A. KNAP. 1994. Carbon system imbalances in the Sargasso Sea. *Nature* **372**, 537–540.
- MORRIS, B., J. BARNES, F. BROWN, AND J. MARKHAM. 1977. The Bermuda marine environment. Bermuda Biological Station Special Publication 15.
- OHDE, S., AND R. VAN WOESIK. 1999. Carbon dioxide flux and metabolic processes of a coral reef. *Okinawa Bull. Mar. Sci.* **65**: 559–576.
- SMITH, S. D. 1988. Coefficients for sea surface wind stress, heat flux, and wind profiles as a function of wind speed and temperature. *J. Geophys. Res.* **93**: 15,467–15,472.
- SMITH, S. V., AND R. W. BUDEMEIER. 1992. Global change and coral reef ecosystems. *Annu. Rev. Ecol. Syst.* **23**: 89–118.
- STEINBERG, D. K., C. A. CARLSON, N. R. BATES, R. J. JOHNSON, A.

- F. MICHAELS, AND A. H. KNAP. 2001. The U.S. JGOFS Bermuda Atlantic Time-series Study: A decade-scale look at ocean biology and biogeochemistry. *Deep-Sea Res. II* **48**: 1405–1447.
- TAKAHASHI, T., J. OLAFASSON, J. G. GODDARD, D. W. CHIPMAN, AND S. C. SUTHERLAND. 1993. Seasonal variation of CO<sub>2</sub> and nutrients in the high-latitude surface oceans: a comparative study. *Global Biogeochemical Cycles* **7**: 843–878.
- , T., R. A. FEELY, R. F. WEISS, R. WANNINKHOF, D. W. CHIPMAN, S. C. SUTHERLAND, AND T. TAKAHASHI. 1997. Global air–sea flux of CO<sub>2</sub>: An estimate based on measurements of sea–air *p*CO<sub>2</sub> difference. *Proc. Natl. Acad. Sci. USA* **94**: 8292–8299.
- WANNINKHOF, R. 1992. Relationship between wind speed and gas exchange over the ocean. *J. Geophys. Res.* **97**: 7373–7382.
- , AND W. R. MCGILLIS. 1999. A cubic relationship between air–sea CO<sub>2</sub> exchange and wind speed. *Geophys. Res. Lett.* **26**: 1889–1892.
- WARE, J. R., S. V. SMITH, AND M. L. REAKA-KUDLA. 1992. Coral reefs: Sources or sinks of atmospheric CO<sub>2</sub>? *Coral Reefs* **11**: 127–130.
- WEISS, R. F. 1974. Carbon dioxide in water and seawater; the solubility of a non-ideal gas. *Mar. Chem.* **2**: 203–215.
- YAMAMURO, M., H. KAYANNE, AND M. MINAGAWA. 1995. Carbon and nitrogen isotopes of primary producers in coral reef ecosystems. *Limnol. Oceanogr.* **40**: 617–621.

*Received: 31 January 2001*

*Accepted: 17 September 2001*

*Amended: 10 October 2001*